Occurrence of N¹⁰-formyltetrahydrofolic acid and its general involvement in transformylation*

 N^{10} -formylfolic acid** has been implicated in one-carbon transfer reactions^{1,2}. This communication presents evidence that the actual compound involved is N^{10} -formyltetrahydrofolic acid or a closely related derivative^{3,4} and that the formyl group is in equilibrium with one-carbon donors and acceptors and with the formyl group of citrovorum factor.

N¹º-CHOFAH₄*** has been isolated and identified as a product of the reaction of formate, ATP and FAH₄ in pig liver extracts (tetrahydrofolate formylase) and of the conversion of serine to glycine after oxidation with DPN\$. It has been isolated by paper chromatography under N₂ in the presence of EDTA and shown to be identical by paper chromatography and absorption spectrum with N¹⁰-CHOFAH₄ synthesized by catalytic hydrogenation⁸ of X¹⁰-CHOFA. Heating of N¹⁰-CHOFAH₄ with o.1 N NaOH anaerobically should yield CF⁰. Treatment of acctone-deproteinized filtrates of the tetrahydrofolate formylase reaction in this manner yielded CF nearly quantitatively based on the formate incorporated. The CF was isolated by paper chromatography. In control reactions without FAH₄ no CF was detected. Interconversion of CF and N¹⁰-CHOFAH₄ has been demonstrated by employing formyl-labeled CF, formyl-labeled N¹⁰-CHOFAH₄, or ring-labeled folic acid and by isolating the folate compounds by chromatography^{§§}. Table I shows that the ¹⁴C-labeled β-carbon of serine is transferred to FAH₄ and that (after an oxidation step) it forms the CHO group of N¹⁰-CHOFAH₄ prior to its appearance in CF.

When ¹⁴C formate was employed as the formyl source, the label again appeared first in N¹⁰-CHOFAH₄ and then in CF⁴.

TABLE I

conversion of β -carbon of serine to N¹⁰-CHOFAH₄ and to CF *

Conditions: 10 μ moles serine-3-¹⁴C, 7,300 cpm./ μ mole, 2 μ moles FAH₄, 1.5 μ moles ATP, 5 μ moles MgCl₂, 2 μ moles MnSO₄, 0.3 ml pig liver extract (1 to 1 homogenate of liver in 0.1 M K-PO₄, pH 7.4, and centrifuged at 27,000 × g for 30 min). Final vol. 1.0 ml, temp. 37 $^{\circ}$, gas N₂.

Time min	Scrine µmoles	N ¹⁰ -CHOFAH ₄ µmoles	CF µmoles	CF = N ¹⁰ - CHOFAH ₄ µmoles
О	o	o	О	O
15	0.013	0.012	O	0.012
30	0.037	0.022	0.006	0.028
45	0.065	0.027	0.049	0.076
60	0.103	0.019	0.073	0.092

^{*} No ¹⁴C formate was found.

TABLE II

Transformylation of the formyl group of $N^{10}\text{-}CHOF\Lambda$ to serine, purine and histidine

Conditions as in Table I, but plus 10 μ moles glycine, 10 μ moles glutamine, 17 μ moles glucose-6-phosphate. 0.8 μ moles DPN, 0.2 μ moles N¹⁰-14CHOFA, 6500 cpm./ μ mole. Time 60 min. In expt. II 2 μ moles of unlabeled HCOOH were added as a pool.

¹⁴ C-substrate	¹⁴ C-compound isolated	I cpm*	II cpm*
N ¹⁰ - ¹⁴ CHO-FA, 1300 cpm.	Serine** Purine*** Histidine Ba-N ¹⁰ -CHOFA BaCO ₃ HCOOH	283 57 34 760 31 6	2.46
	Total	1171	

^{*} Uncorrected for self-absorption.

^{**} β -carbon counted as formdimedon.

^{***} Ag salts of purines in acetone filtrate.

^{*} From a Habilitation Thesis submitted by L. Jaenicke, to the University of Marburg, June, 1954.

^{1954.}
** Abbreviations: FAH_4 , tetrahydrofolic acid; N^{10} -CHOFAH $_4$, N^{10} -formyltetrahydrofolic acid; N^{10} -CHOFA, N^{10} -formylfolic acid; CF, citrovorum factor (N^5 -formyltetrahydrofolic acid); EDTA, ethylenediaminetetraacetate.

ethylenediaminetetraacetate. *** Recently other workers have isolated from bacterial sources compounds with some properties similar to those of N¹⁰-CHOFAH₄5,6,7.

[§] Recent studies have shown that TPN is more active in this system.

^{§§} By ascending chromatography (Schleicher and Schüll 2043 B paper) with 12 $^{\circ}_{00}$ Na₂HPO₄ containing 0.2 $^{\circ}_{00}$ EDTA as a solvent, the R_F values of authentic compounds were: folic acid, 0.24, FAH₄, 0.48, N¹⁰-CHOFA 0.78, N¹⁰-CHOFAH₄, 0.54, CF, 0.62.

While N^{10} -CHOFAH₄ has been shown to occur in these reactions, it is conceivable that it is interconvertible with a more labile derivative. N^{10} -CHOFA is converted to N^{10} -CHOFAH₄ by a reducing system. In Table II 14 C-formyl-labeled N^{10} -CHOFA is shown to transformylate to glycine to form serine. A system forming reduced pyridine nucleotide is required. Transformylation for synthesis of histidine and purine derivatives also is shown.

These experiments have been repeated with labeled N¹¹¹-CHOFAH₄ and gave essentially similar results, but addition of DPNH was still necessary for serine synthesis. Beginning either with serine and FAH₄ or with N¹¹-CHOFAH₄ and DPNH a hydroxymethylfolic acid compound was formed * . The conversion to the hydroxymethyl level by reaction with reduced pyridine nucleotide can be coupled in this system with reduction of DPN to DPNH by a tetrahydrofolate compound. Thus in the presence of substrate amounts of formate, glycine, ATP and DPNH, serine synthesis occurred with catalytic quantities of FAH₄ (0.01 moles/mole serine). However when substrate amounts of FAH₄ were used, only catalytic concentrations of DPNH (0.05 moles/mole serine) were required. While it is clear that a stable cofactor acts in one-carbon transfer reactions³, 7, 10-12, these studies provide evidence that N¹¹-CHOFAH₄ can function as a general transformylating agent, that its formyl group is in equilibrium by an oxidation-reduction system with the β -carbon of serine through a hydroxymethyl level derivative and that CF is not directly involved in this interconversion.

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*On isolation this compound was found to be oxidized to N¹⁰-hydroxymethylfolic acid which was identified by comparison with the authentic compound synthesized by two different methods.

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On the occurrence of N¹⁰-formyltetrahydrofolic acid by enzymic formylation of tetrahydrofolic acid and on the mechanism of this reaction

Pigeon^{1,2} and pig³ liver extracts catalyze an ATP*-dependent reaction between tetrahydrofolic acid and formate to yield a product which transformylates directly to 5-amino-4-imidazolecarbox-amide-5'-phosphoriboside^{1,4} to form inosine-5'-phosphate in the absence of ATP. During purification this compound was converted to a compound having the properties of N¹⁰-formyldihydrofolic acid² and then to N¹⁰-formylfolic acid^{1,2,3,5,6,7}. This communication provides evidence that the overall reaction, catalyzed by an enzyme system which we propose to call tetrahydrofolate formylase, may be formulated as follows:

$$\mathsf{HCOOH} + \mathsf{FAH_4} + \mathsf{ATP} {\longrightarrow} \mathsf{CHOFAH_4} + \mathsf{ADP} + \mathsf{H_3PO_4}$$

The evidence that $CHOFAH_4$ is N^{10} - $CHOFAH_4$ or a closely related compound is: 1. synthetic N^{10} - $CHOFAH_4$, transformylates to imidazolecarboxamide ribotide in the absence of ATP, 2. $CHOFAH_4$ can be converted quantitatively to the N^5 - N^{10} -imidazolinium derivative of formyltetrahydrofolic

^{*}Abbreviations: FAH₄, tetrahydrofolic acid; CHOFAH₄, formyltetrahydrofolic acid; N¹⁰-CHOFAH₄, N¹⁰-formyltetrahydrofolic acid; N¹⁰-CHOFAH₂, N¹⁰-formyldihydrofolic acid; N¹⁰-CHOFA, N¹⁰-formylfolic acid; CF, citrovorum factor; ATP, adenosinetriphosphate; ACF, the N⁵-N¹⁰-imidazolinium derivative of formyltetrahydrofolic acid (anhydrocitrovorum factor, anhydroleucovorin); EDTA, ethylenediaminetetraacetate.